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L. C. R. Andrade,^a* J. A. Paixão,^a M. J. M. de Almeida,^a F. M. Fernandes Roleira,^b M. L. Sá e Melo,^b A. S. Campos Neves^b and E. J. Tavares da Silva^b

^aCEMDRX, Departamento de Física, Faculdade de Ciências e Tecnologia, Universidade de Coimbra, P-3004-516 Coimbra, Portugal, and ^bCentro de Estudos Farmacêuticos, Laboratório de Química Farmacêutica, Faculdade de Farmácia, Universidade de Coimbra, P-3004-295 Coimbra, Portugal

Correspondence e-mail: jap@pollux.fis.uc.pt

Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.003 Å R factor = 0.035 wR factor = 0.108 Data-to-parameter ratio = 8.8

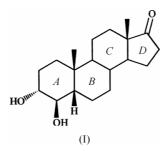
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. The ring conformation of the title compound, $C_{19}H_{30}O_3$, is similar to that of the 5α -epimer except for the *cis A/B* ring junction. Ring *D* adopts a 13β , 14α -half chair conformation. The molecules are linked together by a two-dimensional network of hydrogen bonds involving the carbonyl and hydroxyl groups.

 3α , 4β -Dihydroxy- 5β -androstan-17-one

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Comment

The title compound, $3\alpha, 4\beta$ -dihydroxy- 5β -androstan-17-one, (I), is a 3,4-*trans*-diequatorial diol which has been prepared as a potential intermediate for the synthesis of the potent aromatase inhibitor Formestane (Tavares da Silva *et al.*, 2002), clinically used as an anti-tumor agent in the treatment of estrogen-dependent breast cancers. Preparation of the abovementioned diol results from the abnormal ring opening of the $3\beta, 4\beta$ -epoxide precursor recently studied through X-ray and deuterium-labeling experiments (Tavares da Silva *et al.*, 2002). Following our work on the elucidation of several androstane structures as potential aromatase inhibitors and intermediates in their syntheses, the present X-ray analysis aims to contribute to the elucidation of the reactivity of these compounds.



From the X-ray analysis of (I), a β configuration for this molecule was evident, with a *cis* junction between rings A and B. An ORTEPII (Johnson, 1976) drawing of the molecule with the corresponding atomic numbering scheme is shown in Fig. 1. Bond lengths and angles are within the expected ranges (Allen *et al.*, 1987), with average values: $Csp^3 - Csp^3$ 1.533 (11), $Csp^3 - Csp^2$ 1.519 (4), C=O 1.206 (3) and C-O 1.428 (5) Å. Rings A, B and C have slightly flattened chair conformations, with average torsion angles of 53.7 (9), 55.4 (8) and 55.0 (15)°, respectively. Ring D adopts a 13 β ,14 α -halfchair conformation, with puckering parameters (Cremer & Pople, 1975) calculated using the atom sequence C13,C14,...,C17 of $q_2 = 0.435$ (3) Å and $\varphi_2 = 199.7$ (4)° [pseudo-rotation (Altona *et al.*, 1968) and asymmetry parameters: $\Delta = 2.4$ (4), $\varphi_m = 44.7$ (1), $\Delta C_s(14) = 15.5$ (3) and

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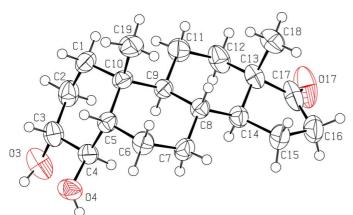


Figure 1

ORTEPII (Johnson, 1976) plot of the title compound. Displacement ellipsoids are drawn at the 50% probability level.

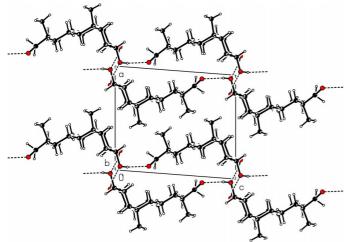


Figure 2

View (Spek, 2003) of the unit cell along the b axis, showing the molecular packing and hydrogen-bond network.

 $\Delta C_2(13,14) = 1.6 (2)^{\circ}$]. The unusual *D*-ring conformation is very similar to that found for the 5α -epimer of this compound (Paixão et al., 1998). The environment around C17 is planar, viz. atoms C13, C16, C17 and O17 are coplanar. The bowing angle between the A ring and the least-squares plane that includes the atoms of the B, C and D rings is $67.62(5)^{\circ}$; the distance between terminal O atoms $(O3 \cdots O17)$ is 8.691 (4) Å and the pseudo-torsion angle C19-C10-C13-C18 is -2.8 (2)°, showing that the molecule is only slightly twisted. The crystal packing is stabilized by a two-dimensional network of hydrogen bonds involving the hydroxyl and carbonyl groups, linking the molecules in planes parallel to (001) (Fig. 2). The 3α -hydroxyl group takes part in two intermolecular hydrogen bonds, one as a donor to the carbonyl group, the other as an acceptor from the 4β -hydroxyl group. The two bonds have approximately the same length; the 3α -hydroxyl to the 4β -hydroxyl intermolecular bond links two molecules head-to-head and the bond via the carbonyl group links two molecules head-to-tail. In addition, there is a short contact between atoms O3 and O4 [2.853 (3) Å, $100 (3)^{\circ}$ at H3] that may correspond to an intramolecular interaction between the two hydroxyl groups.

Experimental

The title compound has been prepared and characterized according to a procedure recently described in the literature (Tavares da Silva et al., 2002). Crystals suitable for X-ray experiments were obtained by slow evaporation of a solution of the steroid in methanol.

 $D_{\rm r} = 1.178 \,{\rm Mg}\,{\rm m}^{-3}$

Cell parameters from 25

Cu K α radiation

reflections $\theta=17.4{-}31.4^\circ$

 $\mu = 0.61 \text{ mm}^{-1}$

T = 293 (2) K

 $R_{\rm int} = 0.017$

 $\theta_{\rm max} = 71.9^{\circ}$

 $h = -12 \rightarrow 12$ $k = 0 \rightarrow 9$

 $l = -14 \rightarrow 14$

3 standard reflections

frequency: 120 min intensity decay: 1.6%

Prism, colourless

 $0.48 \times 0.37 \times 0.37 \text{ mm}$

Crystal data

C19H30O3 $M_r = 306.43$ Monoclinic, P2 a = 9.798 (3) Å b = 7.7742 (11) Åc = 11.372 (4) Å $\beta = 94.36 \ (2)^{\circ}$ $V = 863.7 (4) \text{ Å}^3$ Z = 2

Data collection

Enraf-Nonius MACH-3 diffractometer Profile data from ω -2 θ scans Absorption correction: ψ scan (North et al., 1968) $T_{\min} = 0.713, T_{\max} = 0.798$ 3519 measured reflections 1827 independent reflections 1794 reflections with $I > 2\sigma(I)$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0599P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.035$	+ 0.1172P]
$wR(F^2) = 0.108$	where $P = (F_o^2 + 2F_c^2)/3$
S = 1.09	$(\Delta/\sigma)_{\rm max} < 0.001$
1827 reflections	$\Delta \rho_{\rm max} = 0.24 \ {\rm e} \ {\rm \AA}^{-3}$
208 parameters	$\Delta \rho_{\rm min} = -0.13 \text{ e } \text{\AA}^{-3}$
H atoms treated by a mixture of	Extinction correction: SHELXL97
independent and constrained	Extinction coefficient: 0.0148 (16)
refinement	

Table 1

Hydrogen-bonding geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D{\cdots}A$	$D - \mathbf{H} \cdots A$
O4-H4···O3 ⁱ	0.83 (3)	1.93 (3)	2.751 (3)	168 (3)
$O3-H3$ ··· $O17^{ii}$	0.84 (4)	2.04 (4)	2.792 (3)	149 (4)

All H atoms were refined as riding on their parent atoms using SHELXL97 (Sheldrick, 1997) defaults, except for those of the hydroxyl groups, which had their coordinates freely refined with U_{iso} = $1.2U_{eq}$ of the O atoms. The absolute configuration was not determined from the X-ray data, but it was known from the synthetic route. Friedel pairs were merged, as there was no significant anomalous dispersion.

Data collection: CAD-4 Software (Enraf-Nonius, 1989); cell refinement: CAD-4 Software; data reduction: HELENA (Spek, 1997); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEPII (Johnson, 1976); software used to prepare material for publication: SHELXL97.

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References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). J. Chem. Soc. Perkin Trans. 2, pp. S1–19.
- Altona, C., Geize, H. J. & Romers, C. (1968). Tetrahedron, 24, 13-32.
- Cremer, D. & Pople, J. A. (1975). J. Am. Chem. Soc. 97, 1354–1358.
- Enraf–Nonius (1989). *CAD-4 Software*. Version 5.0. Enraf–Nonius, Delft, The Netherlands.
- Johnson, C. K. (1976). ORTEPII. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.
- North, A. C. T., Phillips, D. C. & Mathews, F. S. (1968). Acta Cryst. A24, 351–359.
- Paixão, J. A., Andrade, L. C. R., de Almeida, M. J. M., Silva, M. R., Tavares da Silva, E. J., Sá e Melo, M. L. & Campos Neves, A. S. (1998). Acta Cryst. C54, 89–91.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Spek, A. L. (1997). HELENA. University of Utrecht, The Netherlands.
- Spek, A. L. (2003). J. Appl. Cryst. 36. In the press.
- Tavares da Silva, E. J., Fernandes Roleira, F. M., Sá e Melo, M. L., Campos Neves, A. S., Paixão, J. A., de Almeida, M. J. M., Costa, M. M. R. R. & Andrade, L. C. R. (2002). *Steroids*, 67/3–4, 311–319.